

Synthesis of differentially protected cyclopentitol: its application towards the stereoselective synthesis of 5-*epi*-calditol

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Abstract—The synthesis of differentially protected cyclopentitol derivative **5** is described using ring-closing metathesis of a diene derived from D-mannose. Subsequent chemical modifications such as catalytic osmylation and chain extension using glycidyl triflate completed the synthesis of 5-*epi*-calditol.

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Cyclopentitols are unique derivatives isolated frequently as subunits of biologically active natural products (Fig. 1). Calditol (**1**) was isolated as a cell wall component of the extreme thermoacidophile *Sulfolobus solfatiricus* (a bacterium present in hot springs and responsible for the production of sulfuric acid from elemental sulfur).¹ Its absolute structure was the subject of speculation for a long time and Sinay and co-workers resolved the issue once and for all by the total synthesis of calditol and its possible isomers.² Trehazolin (**2**) and bacterohopanetriol ethers **3** and **4** are some other natural products, which contain cyclopentitol as a subunit.^{3,4} Samarium iodide mediated intramolecular pinacol coupling of suitably disposed ketoaldehydes seemed to be a popular strategy

in the synthesis **1–4**.^{2,5} Intrigued by the structural similarity between the cyclopentitol moieties of **1–4**, we initiated a programme directed towards the synthesis of these subunits preferably using a common starting point. Herein we report the synthesis of a differentially protected cyclopentitol derivative **5**, its stereo-selective dihydroxylation and conversion of the resulting diol into 5-*epi*-calditol **6**.

The synthesis began with the one carbon Wittig olefination of the known lactol **7** (derived from the crossed aldol reaction between diisopropylidene mannofuranose and formaldehyde in the presence of K₂CO₃ in methanol)⁶ using Ph₃P=CH₂ to procure the olefin **8** in 83%

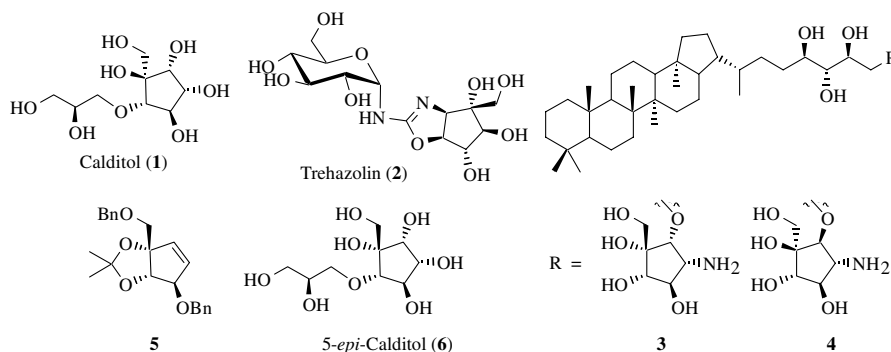


Figure 1.

Keywords: Calditol; Cyclopentitols; Ring closing metathesis; Second generation Grubbs' catalyst; Catalytic osmylation.

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yield. The structure of olefin **8** was confirmed by ^1H NMR spectroscopy, in which signals due to the olefinic protons appeared at 5.32 and 5.92 ppm.

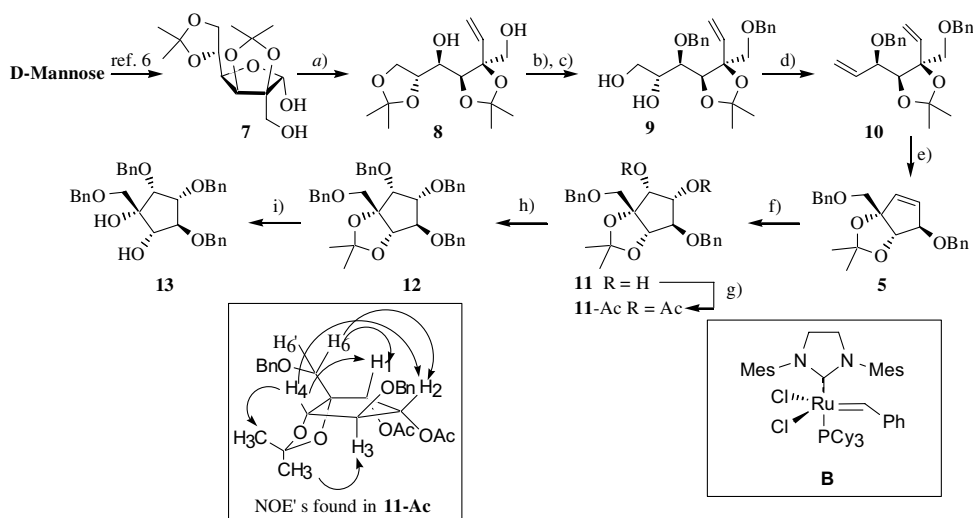
Protection of the diol **8** as the corresponding dibenzyl ether using benzyl bromide and NaH in DMF, followed by selective hydrolysis of the 5,6-isopropylidene group using 0.8% H_2SO_4 in methanol, gave the diol **9** in 82% overall yield. The resulting diol **9** was then treated with methanesulfonyl chloride and triethylamine in CH_2Cl_2 to afford the bis-mesylated compound in quantitative yield, which was subjected to elimination (by treating with sodium iodide in refluxing 2-butanone) without any further purification to afford the diene **10** in 70% yield.⁷ The structure of diene **5** was confirmed by its ^1H NMR spectrum, in which six olefinic proton signals were present between 5.2–5.7 ppm.⁸ As expected, our initial efforts to bring about the ring closing metathesis of diene **10** using the first generation Grubbs' catalyst met with failure. Although, Eustache and co-workers reported the successful RCM of a similar diene using Schrock's catalyst,⁹ however, because of the superior stability as well as easy availability of the Grubbs' second generation catalyst we intended to use it in the present context. Gratifyingly, the RCM of **10** with second generation Grubbs' catalyst (**B**) was facile and the key cyclopentitol derivative **5** was obtained in 81% yield along with the unreacted **10** (11%).¹⁰ The structure of compound **5** was confirmed by its ^1H NMR spectrum, in which only two olefinic proton signals were present at 5.90 (dd, $J = 2.3, 5.8$ Hz) and 6.05 (d, $J = 5.8$ Hz).¹¹

We next focused our attention on the diastereoselectivity of the dihydroxylation of **5**. Eustache and co-workers reported the dihydroxylation of a similar cyclopentitol derivative (albeit protected as its benzyl ethers) with an $\approx 2:1$ diastereomeric ratio, the major being the one in which dihydroxylation occurred from the α -face.⁹ The dihydroxylation of **5** was carried out using osmium

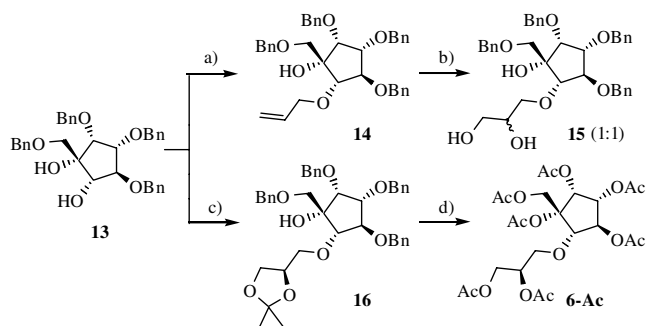
tetroxide and *N*-methylmorpholine *N*-oxide in acetone/water solvent system and occurred exclusively from the α -face giving the diol **11**. For characterization, the diol **11** was converted to its diacetate **11-Ac** by treatment with acetic anhydride in pyridine. In the ^1H NMR spectrum of **11-Ac**, H(1) and H(2) appeared downfield at 5.21 (d, $J = 5.2$ Hz) and 5.29 (dd, $J = 7.3, 5.2$ Hz) ppm, respectively.¹² The observed NOE's between H(1) and H(6)/H(6'), H(4) and H(1)/H(2) in the NOESY spectrum of **11-Ac** clearly indicated a *syn*-spacial relation between these protons thus confirming its assigned structure (Scheme 1). After confirming the configuration of the resulting diol **11**, we proceeded to accomplish the synthesis of 5-*epi*-calditol. Treatment of the diol **11** with NaH and benzyl bromide in DMF followed by deprotection of the acetone group with 80% aq acetic acid afforded **13** in 76% yield.

Our intended strategy for the installation of the glycidyl ether unit was the selective allylation of HO-C(4) of **13** and the Sharpless asymmetric dihydroxylation of resulting allyl ether. Selective allylation of **13** was carried out by treatment with 1.2 equiv each of NaH and allyl bromide in THF, the allyl ether **14** was obtained in 76% yield. Attempted asymmetric dihydroxylation of **14** using ADMix- β gave poor diastereoselectivity (inseparable 1:1 epimeric mixture).¹³ After being met with discouraging results in the asymmetric dihydroxylation, we next focused our attention on alkylation of the diol **13** directly with the corresponding glycerol derivative (Scheme 2).

After initial investigations with different 2,3-*O*-isopropylidene (*R*)-glyceryl halides as well as with the corresponding 1-*O*-sulfonates, we found that **13** could be selectively alkylated (63% yield) by using 2,3-*O*-isopropylidene-(*R*)-glycerol-1-*O*-triflate and *n*-BuLi in THF. Deprotection of 5-*epi*-calditol **16** followed by peracetylation using acetic anhydride in pyridine and catalytic



Scheme 1. Reagents and conditions: (a) $\text{PPh}_3\text{CH}_3^+\text{I}^-$, NaNH_2 , THF/ EtO_2 , 0°C –rt, 10 h, 83%; (b) NaH, BnBr, DMF, 0°C –rt, 4 h, 90%; (c) 0.8% w/v $\text{H}_2\text{SO}_4/\text{H}_2\text{O}$, MeOH, rt, 6 h, 91%; (d) MsCl, Et_3N , CH_2Cl_2 , rt, 4 h, then NaI, 2-butanone, reflux, 6 h, 70% (overall); (e) **B** (2 mol%), C_6H_6 , reflux, 6 h, 81%; (f) OsO_4 , NMO, acetone/ H_2O , rt, 12 h, 80%; (g) Ac_2O , pyridine, DMAP (cat), rt, 4 h, 89%; (h) NaH, BnBr, DMF, 0°C –rt, 6 h, 95%; (i) 80% acetic acid/ H_2O , 60°C , 4 h, 80%.



Scheme 2. Reagents and conditions: (a) NaH, allyl bromide, THF, 0°C–rt, 2 h, 76%; (b) ADmix-β, *t*-BuOH/H₂O, 0°C, 18 h, 78%; (c) *n*-BuLi, 2,3-*O*-isopropylidene-(*R*)-glycerol-1-*O*-triflate, THF, -78–0°C, 6 h, 63%; (d) Pd/C, H₂, MeOH/AcOH (1:0.01), 6 h, then Ac₂O, pyridine, DMAP (cat), 12 h, 61% (overall).

dimethylaminopyridine gave the hepta-acetyl-5-*epi*-calditol **6-Ac** in 61% overall yield. The spectral and analytical properties as well as the optical rotation of **6-Ac** are comparable with the reported values.¹⁴

To conclude, we have provided a simple preparative procedure for the synthesis of differentially protected cyclopentitol **5**, and its application towards the diastereoselective synthesis of 5-*epi*-calditol. Presently, work towards the synthesis of naturally occurring aminocyclopentitol derivatives from **5** using the Overman aza-Claisen rearrangement or epoxidation and subsequent inter/intramolecular opening with amino-nucleophiles is in progress.

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- Spectral data of diacetate **11-Ac**: $[\alpha]_D^{25} +47.6$ (*c* 1.5, CHCl₃); ¹H NMR (CDCl₃, 500 MHz): δ 1.35–1.47 (2s, 6H), 2.04–2.09 (2s, 6H), 3.60 (d, *J* = 10.6 Hz, 1H), 3.61 (d, *J* = 10.6 Hz, 1H), 4.12 (dd, *J* = 2.9, 7.3 Hz, 1H), 4.45 (d, *J* = 2.9 Hz, 1H), 4.58 (d, *J* = 12.1 Hz, 1H), 4.59 (d, *J* = 12.0 Hz, 1H), 4.63 (d, *J* = 12.1 Hz, 1H), 4.70 (d, *J* = 12.0 Hz, 1H), 5.21 (d, *J* = 5.2 Hz, 1H), 5.29 (dd, *J* = 5.2, 7.3 Hz, 1H), 7.25–7.34 (m, 10H); ¹³C NMR (CDCl₃, 125 MHz): δ 20.7, 20.8, 27.1, 27.4, 72.0, 72.9, 73.1, 73.9, 75.6, 85.0, 85.1, 86.6, 114.1, 127.6, 127.7, 127.8, 128.4, 128.5, 137.6, 137.8, 169.3, 169.5. Anal. Calcd for C₂₇H₃₂O₈: C, 66.93, H, 6.66. Found: C, 66.56, H, 6.42.
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